La chimiométrie aux frontières du domaine proche infrarouge : explorer sans a priori les données spectroscopiques Raman, moyen infrarouge et térahertz.

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We need tools !

Modern spectroscopic instrumentation: generation of an always larger amounts of data fairly automatically in a very short period of time.



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The complexity of the data is always higher but what we believe to be an obstacle is a real opportunity to solve the given analytical problem.

development of chemometrics tools/methods to make sense of such data set and extract the maximum amount of useful information.

What we can do with chemometrics ?

- Data pre-processing : filtering, denoising, baseline correction, signal amplification, artifact correction, data conditionning ...
- Design of Experiments (DOE) : a better understanding of the effects that different parameters have on a response (Screening, Saving time, Quantitative modelling, Optimisation)
- Exploratory Data Analysis : looking at relationships between samples / variables. Find structure, find outliers.
- Classification : check for groupings = unsupervised, existing groupings = supervised, visualize groupings, classify.
- Regression and multivariate calibration : relate instrumental measurements to characteristics of the sample of interest.
- Multivariate curve resolution : extract valuable profiles of pure species from multivariate dataset

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Outline

- Probing the hydration shell of molecules with Terahertz Spectroscopy
- Characterizing a single cancer cell with synchrotron infrared microspectroscopy
- In situ monitoring of polymorphic transformations with multiset image analysis for hyperspectral Raman signal unmixing

The potential of Multivariate Curve Resolution methods

Multivariate curve resolution has become one of the most important chemometrics method mainly due to its great potential for data analysis and its adaptability.

Objective : "Recovery of the response profile of pure components in an unresolved and unknown mixture obtained from evolutionary processes".

Roma Tauler

Response profile : spectra, pH profiles, time profiles, elution profiles ...

(*) R. Tauler, Chemomet. Intel. Lab. Systems, 1995, 30, 133-146.

The importance of hydration

Understanding the hydration mechanisms is crucial

Involved in many biological phenomena

Examples:

- Solvated-proteins dynamics is highly correlated to the solvent dynamics.
- Protein hydration plays an important role for the protein function expression

The classical hydration model

Non-charged molecule solvated in water : hydration is made through weak bonds such as hydrogen ones between the molecule and water molecules.

Consequence: the behavior of water molecules at the vicinity of the solvated molecule is changed.

Directly probing hydration shell of molecules

Observation: many studies on this topic but many questions remain (hydration-shell extent, hydration-process time scales or its thermodynamic-parameter dependence).

Main idea: probe directly the hydrogen bond network in water with Terahertz spectroscopy (100GHz-10THz) and obtain additional knowledge on hydration shell structure.

Overall objective: development of a global methodology (spectroscopic acquisition and chemometrics) in order to probe hydration shell of molecules.

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The Terahertz spectral domain

The Terahertz (Thz) spectral domain (teraherzt gap): a good way to probe low energy bonds such as hydrogen ones.

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Problem : THz measurements on liquid samples are not easy due to huge water absorption.

To perform transmission measurements : powerful sources have to be used (with the risk of sample heating) or sample volumes have to be reduced and well-controlled.

Our solution: development of an original microfluidic THz sensor to characterize hydration of molecules in a chip.

Thz spectral acquisition in a microfluidic system

Liquid-absorption measurements: development of a two-function microsystem

- a microfluidic circuit drives the samples towards the analysis area.
- an integrated electromagnetic element (Goubau line) guides the THz waves through the sample to analyze.

Vectorial Network Analyzer connected to the wave-guide: measurement of amplitude and phase of both reflected and transmitted signals through the sample calculation of an absorbance spectrum on the whole spectral domain (like in the mid infrared spectral range).

S. Laurette, A. Treizebre and B. Bocquet, Journal of Micromechanics and Microengineering, 2011, 21, 065029.

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Probing hydration shell of ethanol

Feasibility study of the concept: analyze of the ethanol / water system (thought to be a simple one).

Preparation of aqueous solutions of ethanol (alcohol volume ratio from 0 to 1)

Pure water Water 95%. Ethanol 5% Water 90%, Ethanol 10% Water 85%, Ethanol 15% Water 80%, Ethanol 20% Water 75%. Ethanol 25% Water 70%, Ethanol 30% Water 65%, Ethanol 35% Water 55%, Ethanol 45% Water 45%, Ethanol 55% Water 40%, Ethanol 60% Water 35%. Ethanol 65% Water 30%, Ethanol 70% Water 20%, Ethanol 80% Water 15%, Ethanol 85% Water 10%, Ethanol 90% Water 5%, Ethanol 95% Ethanol 100%

First derivative to suppress a very important baseline shift completely masking the small spectral features.

Using univariate spectral data observation

Possible to observe various signal shapes. Can be even reassuring compared to the classical hydration model.

Univariate observations can't give us an idea about the real number of contributions present in the chemical system. No guarantee that the used frequency is effectively specific to only one compound in the system (responses are often the addition of several contributions).

Univariate spectral data analysis : a partial vision of the chemical system and very often a biased one.

Our goal: use Multivariate Curve Resolution in order to retrieve information about the hydration shell in the ethanol / water system using the whole spectral domain with no a priori.

Applying multivariate curve resolution method

The MCR-ALS^(*) method:

- A model-free approach
- Unravels the **pure contributions** of all species in the spectroscopic dataset **D** without requiring chemical information about the underlying physicochemical system.

Constrained bilinear decomposition of the total instrumental response of a system **D** (*nxf*) into the product of two simpler matrices **C** (*nxp*) and **S**^t (*pxf*) i.e. the concentration profiles and corresponding pure spectra

 $D(n \times f)$: Spectral data matrix, n = number of ethanol / water mixtures, f = frequencies in the spectral domain. $C(n \times p)$: Matrix of pure concentration profiles, n = number of ethanol / water mixtures, p = mathematical rank of D matrix = number of pure contributions in the chemical system.

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- S^t(*p x f*) : Matrix of pure spectral profiles.
- E(n x f) : Residual matrix (unmodelled variations)

(*) R. Tauler, Chemomet. Intel. Lab. Systems, 1995, 30, 133-146.

Evaluate the number of pure contributions

Singular Value Decomposition (SVD) of D

First step of the MCR-ALS methodology :

Evaluate p the mathematical rank of matrix **D** i.e. the number of pure contributions in the system (not an easy task even for a mathematician \bigcirc).

Solution: try to separate significant singular values corresponding to real spectral contributions from non-significant ones corresponding to noise.

Due to a change of slope on the singular value curve, it is possible to estimate a threshold (dotted line) above which the contributions are considered significant.

As an initial assessment, we can say that **four species** are present in our **D** matrix. This first result is very important since it indicates that our microsystem is able to see more than water and ethanol.

Observations: green curve may correspond to « bulk water » and the red one to ethanol. First assumption: blue curve can be the hydration shell contribution.

Inconsistencies: two contributions for pure ethanol, unmodelled spectral variations remain on E

Observations:

- More consistent results with four contributions (also given by SVD).
- Two species (in light and dark blue) appear to be the "non-bulk" water or the hydration shell.

Multivariate Curve Resolution in imaging spectroscopy

Imaging Spectroscopy

- A key field of analytical chemistry.
- Powerful tool for characterizing the molecular distribution of different chemical compounds in heterogeneous materials.
- Development of many chemometrics methods in order to extract more hidden information from such experiments.

Multivariate Curve Resolution in imaging spectroscopy The classical method for concentration maps generation Microscope Acquisition of the spectral data cube 1) Spectrometer (ex: mapping). Selection of a specific wavelength of 2) the compound of interest. x pixels 3) Signal integration to generate the v pitels corresponding chemical map (i.e. a osorbanc slice of the data cube). Sam **Experimental** data cube λ khanee Often we forget that wavelength specificity is a strong hypothesis! LASI

Multivariate Curve Resolution in imaging spectroscopy

Wavelength specificity is a strong hypothesis

Some drawbacks:

• Usually necessary to know all compounds present in the analyzed sample.

If a compound is forgotten = possible interferences = overestimations of concentration = false distribution maps = biased vision of analytical reality.

➢impossible to select a specific spectral range for an unexpected compound (no image).

• Sometimes impossible to find a specific spectral range for each compound due to the complexity of the sample (high number of species) and/or the high bandwidth of the considered spectroscopy.

What can be done with Multivariate Curve Resolution ?

Multivariate Curve Resolution in imaging spectroscopy MCR-ALS Methodology Multivariate curve resolution (bilinear Unfolding of the experimental data cube. decomposition) Rank evaluation of the **D** matrix (*nc*: number Simultaneous extraction with no prior knowledge of spectral contributions in the dataset). Simultaneous extraction of C and St ST n Refolding **C** in order to **Spectral data Pure compounds Pure compounds** generate chemical maps. concentration matrix Spectral matrix Λ matrix Spectral data nc unfolding > = × ×× λ First pure spectra **Experimental** (molecular data cube characterization) **Refolding of the first pure** compound concentration distribution LASI (pixels space)

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MIR Imaging on a single HeLa cancer cell

Why do we need Infrared Synchrotron radiation ?

- 1000 times brighter than our IR spectrometer sources : highest signal to noise ratio -> possible to observe small absorbance variations.
- Possible to obtain the highest spatial resolution : necessary if we want to observe details for very small sample.

Is it really enough?

Rayleigh criterion: $R \sim \lambda$ IR spectral range : 2.5 µm < λ < 25 µm Cell dimensions : few tens micron *What can be done with chemometrics (MCR-ALS and Super-resolution)*

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A first spectral exploration

Characterizing a single HeLa cell with IR Synchrotron microspectrometry

A first spectral exploration

Chemical map extraction with MCR-ALS

Pure spectral contribution #1

The super-resolution concept in spectroscopic imaging

The super-resolution concept

Background: Real discipline of the signal processing community (early works in 1980, a real emerging scientific research in 2000).

Concept: Simultaneous exploitation of several low resolution images of the same object (observed from "different angles") in order to obtain one higher resolution image.

Condition of application: sub-pixellic shift between low resolution images (image shifts lower than the pixel size of low resolution image).

A super-resolution example in space observation

NASA's Viking Mission: the mission objectives were to obtain high resolution images of the Martian surface, characterize the structure and composition of the atmosphere and surface, and search for evidence of life.

Simultaneous use of 24 shifted images of mars acquired by vicking orbiter 1

Visible images

742 m / pixel

Super-resolution

¹⁸⁶ m / pixel

Super-resolution acquition setup

Multiset MCR-ALS analysis + Super-resolution

Analytical Chemistry, 85(13), p6303 (2013).

Polymorphism study with in situ hyperspectral Raman imaging

Polymorphism is defined by the existence of several crystalline forms for the same chemical compound.

Differences in crystal structure = variations in chemical and physical properties = different behavior of the molecule.

In pharmaceutical industry: relevant to know the potential presence of polymorph transformations induced by different parameters, such as light exposure, pressure, temperature changes...

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Aim of the study: use of multiset analysis on the series of in situ Raman hyperspectral images acquired during a thermal induced transformation of carbamazepine as the optimal way to extract useful information about polymorphic transformations.

S. Piqueras, L. Duponchel, R. Tauler, A. de Juan, Analytica Chimica Acta, 819, p15 (2014)

Polymorphism study with in situ hyperspectral Raman imaging

Experimental setup

Sample: anhydrous commercial carbamazepine (CBZ) obtained from Sigma–Aldrich (≥99% r.a). Powder slightly compacted (2 mm thickness) in order to obtain a flat surface.

Raman imaging: Labram HR spectrometer, Horiba JY (@632.81nm, from 100 to 1650cm–1). Integration time 0.5 s, pixel size 10 μ , mapping size 20 x 20 pixels. Total acquisition time / cube: 40 min.

Nine Raman data cubes were collected along the thermal degradation at different temperatures (from T=25 \circ C to T = 160 \circ C with THMS600 accessory from Linkam).

Next step: global rank estimation

The global rank is potentially an estimation of the total number of compound in the system

Singular value decomposition (SVD) applied on the columnwise augmented matrix **D**

Detection of three significant singular values = three spectral contributions present in the system.

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Exploratory local rank analysis

Provide information on the local complexity (compound overlap) at the pixel level of the sample analyzed

Given an experimental data cube, the FSIW-EFA algorithm (Fixed-Size image Window -Evolving Factor Analysis) performs PCA analyses by moving small windows around each individual pixel area across the full image (*).

- @ 25 °C: zones with 2 compounds are observed (rather unexpected).
- increase of the complexity up to 130 °C (more green regions).
- Image complexity decreases until 160 °C (where only some pixels with 2 compounds).

The more important: observing pixels in the image where the number of overlapping constituents is smaller than the total (i.e., where some constituents are absent) is crucial to obtain a unique solution in the image multiset resolution analysis.

Image multiset MCR-ALS analysis

Spectral interpretation

- Possible to retrieve the two well-known polymorph forms 1 and 3 also called alpha form and beta form respectively.
- Extraction of a unknown polymorph C with specific spectral features.
- Difficult to have a more important spectral overlap between pure compounds: more or less impossible to generate unbiased chemical maps with the classical integration method.

Chemical maps interpretation

- Distribution maps allow following the process evolution at a pixel (local) level and interpret spatial process evolution.
- Polymorphs B and C are present at 25°C.
- Polymorph A emerges and B and C decay as temperature increases along the process.
- All compounds melt at the end of the process with the consequent decrease in Raman signal and loss of intensity in the distribution maps.

Chemical maps interpretation

- Observation of different dominant spatial zones of initial compounds B and C.
- Polymorphic product A extends on the spatial areas of B and C
 => the single final product can be obtained from the two initial ones following the parallel pathways :

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Mean concentrations of **B** and **C** are correlated along the process

 \Rightarrow if we don't use imaging spectroscopy but a classical one for bulk analysis, MCR would extract only one contribution for the initial product (probably a mix of B and C).

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 \Rightarrow Impossible to discover such the parallel pathways.

- Richness of vibrational spectrscopy
- Chemometrics is a good way to push the limits of scientific instrumentation (high level or routine basis)
- Chemometrics and more precisely Multivariate Curve Resolution method offers significant potential for the analysis of complex dataset with no prior knowledge about the underlying physico-chemical system. A High degree of adaptability.
- Chemometrics is accessible to all (I'm just a physical-chemist ⁽ⁱ⁾) as long as we take the time to discover it.
- Chemometrics is not a push button concept!

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Thank you for your attention!

